

## Chapter 11 Supplemental Problems Stoichiometry Answers

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370 Chapter 11 • Stoichiometry EXAMPLE Problem 11.1 Interpreting Chemical Equations The combustion of propane (  $C_3H_8$  ) provides energy for heating homes, cooking food, and soldering

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metal parts. Interpret the equation for the combustion of propane in terms of representative particles, moles, and mass.

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We can see from the stoichiometry of the reaction that  $\frac{3}{2}$  mol of  $O_2$  is required to produce 1 mol of  $H_2SO_4$ . This is a standard stoichiometry problem of the type presented in Section 11.4, except this problem asks for the volume of one of the reactants ( $O_2$ ) rather than its mass. We proceed exactly as in Section 11.4, using the strategy

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Be able to identify and write balanced chemical equations to solve stoichiometry problems

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